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## ITEM 3

A casual worker at Namungona sewage treatment plant was given a pack of pellets of substance W to be used to treat sewage. On the pack, it is noted that every  $500 \text{cm}^3$  of sewage requires 0.67moles of W to be purified. On the same pack, it is also indicated that every 2.6g of gaseous substance W occupies 1.7577 litres of space at STP. Further still, the total weight of the pack was noted as 50,000g.

Among the cautions on the pack, was; "do not expose substance W to any calcium containing substance" as it reacts as follows. " $Ca_{(s)} + W_{2(s)} \longrightarrow CaW_{2(s)}$ . However, she failed to estimate the volume of sewage that required a single pack of W. She has approached you for help.

## Task.

As a chemistry learner with enough knowledge, make a presentation with

- a) An explanation of the category of a substance W is.
- b) The guidelines the casual worker can use to treat the sewage.
- c) Challenges associated with repeated use of such substances to the environment.

## (1moles of a gas occupies 24ltrs at STP)